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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.		
10/024,243 12/21/2001		Yoshiro Shiokawa	111522	3419		
25944	7590 12/19/2003		EXAM	EXAMINER		
OLIFF & BERRIDGE, PLC			JOHNSTON, PHILLIP A			
P.O. BOX 19928 ALEXANDRIA, VA 22320			ART UNIT	ART UNIT PAPER NUMBER		
		•	2881			

DATE MAILED: 12/19/2003

Please find below and/or attached an Office communication concerning this application or proceeding.

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	Applica	ntion No.	Applicant(s)					
Office Action Summary		,243	SHIOKAWA ET A	SHIOKAWA ET AL.				
		ier	Art Unit					
		Johnston	2881					
The MAILING DATE of this communication appears on the cover sheet with the correspondence address Period for Reply								
A SHORTENED STATUTORY PERIO THE MAILING DATE OF THIS COMM Extensions of time may be available under the provi after SIX (6) MONTHS from the mailing date of this If the period for reply specified above is less than thi If NO period for reply is specified above, the maximu Failure to reply within the set or extended period for Any reply received by the Office later than three more earned patent term adjustment. See 37 CFR 1.704(Status	UNICATION. sions of 37 CFR 1.136(a). In no communication. irty (30) days, a reply within the s um statutory period will apply and reply will, by statute, cause the a nths after the mailing date of this	event, however, may a repl tatutory minimum of thirty (d will expire SIX (6) MONTH pplication to become ABAN	ly be timely filed 30) days will be considered time IS from the mailing date of this of NDONED (35 U.S.C. § 133).	ely. communication.				
1) Responsive to communication(s) filed on							
2a)⊠ This action is FINAL .	2b) ☐ This action is	non-final.						
3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.								
Disposition of Claims								
4) Claim(s) <u>1-10</u> is/are pending in the day Of the above claim(s) 5) Claim(s) is/are allowed. 6) Claim(s) <u>1-10</u> is/are rejected. 7) Claim(s) <u>1-8</u> is/are objected to. 8) Claim(s) are subject to results.	is/are withdrawn from o							
Application Papers		•						
 9) The specification is objected to be 10) The drawing(s) filed on 21 December 10. Applicant may not request that any of the control of	<u>nber 2001</u> is/are: a)⊠		•	niner.				
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).								
11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.								
Priority under 35 U.S.C. §§ 119 and 120								
12) Acknowledgment is made of a classified copies of the prious of the prious of the copies of the prious of the copies of the prious of the certified copies of the priod of the priod of the certified copies of the priod of the priod of the priod of the priod of the certified copies of the priod of the pri	of: rity documents have be rity documents have be rity documents have be ies of the priority docum ational Bureau (PCT Ri ction for a list of the cei m for domestic priority uded in the first sentend language provisional a m for domestic priority	een received. een received in App nents have been re ule 17.2(a)). rtified copies not re under 35 U.S.C. § ce of the specification application has bee under 35 U.S.C. §§	elication No ceived in this National ceived. 119(e) (to a provisiona on or in an Application n received. § 120 and/or 121 since	al application) Data Sheet. a specific				
Attachment(s)		5-7						
 Notice of References Cited (PTO-892) Notice of Draftsperson's Patent Drawing Revie Information Disclosure Statement(s) (PTO-144 	•	-	nmary (PTO-413) Paper No(rmal Patent Application (PT0					

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Detailed Action

Examiners Response to Arguments

1. The amendment filed 10-06-2003 is objected to under 35 U.S.C. 132 because it introduces new matter into the disclosure. 35 U.S.C. 132 states, that no amendment shall introduce new matter into the disclosure of the invention. The added material, which is not supported by the original disclosure is as follows: "third body gas".

Applicant is required to cancel the new matter in the reply to this Office Action.

Claims Rejection – 35 U.S.C. 103

- 2. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
- (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which the subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 3. Claims 1-10 as amended are rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. Patent No. 6,566,652 to Kato, in view of Kato, U.S. Patent No. 6,008,490 and in further view of Sato, U.S. Patent No. 5,194,739, for the reasons given in First Office Action (Paper No. 4).

4. Applicant's arguments filed 10-06-2003 have been fully considered but they are not persuasive.

Arguments 1 and 2.

Applicant states that "None of the applied art teaches, discloses or suggests ion attachment mass spectrometry for attaching positively charged metal ions emitted from an ion emitter to a gas to be detected, as claimed in Claim 1 and similarly claimed in claims 2-8."

Applicant also states that "In Kato, the mass spectrometry has no ion emitter 18 in the ion source."

The applicant is respectfully directed to Kato (652), Column 5, line 39-67, which states; In the atmospheric pressure ion source 7, a bath gas to regularize flow of the gas in the atmospheric pressure ion source 7, and to prevent contamination inside the cover case 8, and a counter gas which promotes evaporation of the nebulized droplets, are passed, according to necessity. For these gases, needle valves control the quantity of the gas flow independently and respective independent stop valves for switching the flows on/off are provided.

A direct current high voltage of about 3 to 6 kV supplied from a high voltage power supply (not shown in the figure) is applied to a tip edge of the nebulizer probe 4. The solution is nebulized from the tip of the probe 4 into the atmosphere of the atmospheric pressure ion source 7 as fine droplets 6 respectively having charges due to the high electric field generated in the tip neighborhood of the probe 4 by this high

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voltage and a nitrogen gas for nebulization. The fine droplets 6 having charges collide with the neblization gas molecules while flying in the atmosphere so that solvent at the surface of the droplet is vaporized. Therefore, the sample ions contained in the droplets finally are expelled (emitted) into the atmosphere in the atmospheric pressure ion source 7. The ions which are generated, enter into the high vacuum part 16 evacuated by the vacuum pump 22 as an ion beam 17, from the aperture 11 for sampling the ions, through the intermediate pressure chamber 12 evacuated by the vacuum pump 23 and the aperture 14. The ions travel to the mass spectrometer 15, are mass-analyzed there, and are detected by the detector 18 so as to provide a mass spectrum or a mass chromatogram using the data processor 19.

Also in Sato (739) Column 7, line 22-25, which states; A liquid metal ion source which can produce cesium ions stably for a long time in the form of a beam focussed to a micro-spot. The liquid metal ion source is composed of a reservoir containing a liquid metal, and a needle type emitter passing through the reservoir and having a sharp tip end which protrudes from the reservoir, the liquid metal being composed primarily of a cesium compound containing 0.3-20 atom % of oxygen.

The examiner has interpreted from the Kato (652) and Sato (739) references above that metal droplets are emitted into an ion source in the presence of a high voltage field, which then collide with a sample gas facilitating ion attachment, as recited in the independent Claims 1-8.

Conclusion

5. The Amendment filed on 10-06-2003 under 37 CFR 1.131 has been considered but is ineffective to overcome the Kato (652) and Sato (739) references.

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

5. Any inquiry concerning this communication or earlier communications from the examiner should be directed to Phillip A Johnston whose telephone number is 305 7022. The examiner can normally be reached on 7:30 to 4:00.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, John R Lee can be reached on 703 308 4116. The fax phone numbers for the organization where this application or proceeding is assigned are 703 872 9318 for regular communications and 703 872 9319 for After Final communications.

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Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is 703 308 0956.

ΡJ

December 15, 2003

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